Synthesis and Biological Activities of N-Phosphoryl Branched Peptides

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Abstract: A series of N-phosphoryl branched peptides were synthesized by coupling of various N-phosphoryl amino acids to L-Lysine methyl ester, and their structures were confirmed by ³¹P NMR, ¹H NMR, MS and elemental analysis. The results of cell biological tests indicated that compound **1d** and **1e** obviously inhibited the growth of both K562 and A2780 cells.

Keywords: N-Phosphoryl branched peptide, synthesis, biological activity, apoptosis.

Much research has shown that some N-phosphoryl peptides are endowed with important biological activities. For instance, N^{α} -(diaryloxyphosphoryl)-L-alanyl-L-prolines are moderate inhibitors of angiotensin converting enzyme and yield highly potent inhibitors when they hydrolyze under physiological condition lossing 1 mol phenol¹; Phosphoramido (N-[(\alpha-rhamnopyranosyloxy)hydroxyphosphinyl]-L-Leu-L-Trp) and its analogues have been shown to be inhibitors of endothelin converting enzyme, therefore, they may be used in the therapies of hypertension, acute renal failure, and coronary or cerebral vasospasm caused by overprodction of endothelin^{2,3}. Dendritic branched peptides synthesized with L-lysine or its oligopolymer as matrix⁴ have been widely used in the research and development of immunodiagnostic reagent⁵ and synthetic peptide vaccine⁶. Furthermore, it was found that N-phosphoryl amino acids have some special properties⁷. Therefore, we synthesized a series of N-phosphoryl branched peptides by coupling of various N-phosphoryl amino acids to L-lysine methyl ester. All their structures were confirmed by ³¹P NMR, ¹H NMR, MS and elemental analysis (Table 1, 2), and preliminary cell biological tests indicated that two of them obviously inhibited the growth of both K562 and A2780 cells.

In the literature^{1,3}, N-protected peptides were firstly prepared by conventional method, then phosphorylation of the deprotected products yielded N-phosphoryl peptides. In this paper, coupling of various N-phosphoryl amino acids prepared according to the method established by our laboratory⁸ to L-lysine methyl ester gave N-phosphoryl branched peptides in good yields, thereby the steps of protection of amino acids and

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deprotection of peptides have been eliminated. Furthermore, one of the N-phosphoryl branched peptide esters has been converted into the corresponding peptide amide and peptide acid (**Scheme 1**).



General procedure for the preparation of compounds 1a - h: L-Lysine methyl ester dihydrochloride (2 mmol), N-methylmorpholine (4 mmol), N-phosphoryl amino acid (4.2 mmol) and 1-hydroxybenzotrizole (4.2 mmol) were dissolved in dry tetrahydrofuran (10 mL), the solution was stirred and cooled in an ice-water bath while dicyclohexylcarbodiimide (4.2 mmol) in THF (5 mL) was added dropwise. After stirring overnight at room temperature, the precipitate was removed by filtration and the solvent evaporated to dryness *in vacuo*. A mixture of a saturated solution of NaHCO₃ and CHCl₃ was added to the residue and the organic phase washed with saturated NaHCO₃, 10% citric acid, and saturated NaCl aqueous solution. The solution was dried over anhydrous Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The residue was purified by chromatography on a column of silica gel with CHCl₃-MeOH (9:1) as eluent, or recrystallization from THF-petroleum ether to yield compounds **1a~h**.

 Table 1
 Physical constants and elemental analysis data of compounds 1, 2 and 3

Compd.	AA	R	Yield	m. p.	Formula	Elemental analysis (%, Calcd.)		
			(%)	(°C)		C	H	N
1a	Ala	i-Pro	77.0	69-70	$C_{25}H_{52}N_4O_{10}P_2$	47.58(47.62)	8.39(8.25)	8.91(8.89)
1b	Ala	Et	68.7	92-94	$C_{21}H_{44}N_4O_{10}P_2 \cdot H_2O$	42.55(42.57)	7.55(7.78)	9.59(9.46)
1c	Ile	i-Pro	71.3	163-164	$C_{31}H_{64}N_4O_{10}P_2$	52.45(52.10)	9.25(8.96)	7.90(7.84)
1d	Leu	i-Pro	81.3	115-117	$C_{31}H_{64}N_4O_{10}P_2$	52.29(52.10)	9.20(8.96)	7.89(7.84)
1e	Phe	i-Pro	55.1	106-108	$C_{37}H_{60}N_4O_{10}P_2$	56.85(56.78)	7.60(7.67)	7.28(7.16)
1f	Pro	i-Pro	93.1	oil	$C_{29}H_{56}N_4O_{10}P_2\cdot^3/_2H_2O$	49.01(49.08)	8.02(8.18)	8.08(7.90)
1g	Ser	i-Pro	74.0	86-88	$C_{25}H_{52}N_4O_{12}P_2$	45.07(45.32)	7.81(7.85)	8.19(8.46)
1h	Thr	i-Pro	70.3	136.5-138	$C_{27}H_{56}N_4O_{12}P_2$	46.80(46.96)	8.11(8.12)	7.85(8.12)
1i	Val	i-Pro	75.9	160-162	$C_{29}H_{60}N_4O_{10}P_2$	50.98(50.73)	8.78(8.75)	8.19(8.16)
2	Ala	i-Pro	61.2	127-128	$C_{24}H_{51}N_5O_9P_2 \cdot 1/2H_2O$	45.93(46.15)	8.09(8.33)	10.95(11.22)
3	Ala	i-Pro	71.9	55-56	$C_{24}H_{50}N_4O_{10}P_2 \cdot H_2O$	45.66(45.42)	8.16(8.20)	8.93(8.83)

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Compd	ESI-MS	³¹ P NMR	¹ H NMR $(\delta \text{ ppm})^{**}$		
compu.	(m/z)	(δ, ppm)			
1a	631*	6.43	1.29-1.39(m, 24H, CH ₃ ×8), 1.41-1.46(m, 6H, Ala CH ₃ ×2), 1.50-1.58(m, 4H, Lys		
	$[M+H]^+$		γ -CH ₂ and δ -CH ₂), 1.57-1.95(m, 2H, Lys β -CH ₂), 3.29(t, 2H, Lys ϵ -CH ₂), 3.75(s,		
			3H, OCH ₃), 3.75(m, 2H, Ala α -CH×2), 4.40(m, 1H, Lys α -CH), 4.60(m, 6H,		
			OCH×4 and Ala NH×2), 7.16(t, 1H, Lys ε -NH), 7.29(d, 1H, Lys α -NH)		
1b	575.2	8.34	1.31(t, 12H, CH ₃ ×4), 1.36-1.43(m, 8H, Ala CH ₃ ×2 and Lys γ -CH ₂), 1.55(m, 2H,		
	[M+H] ⁺		Lys δ -CH ₂), 1.81(m, 2H, Lys β -CH ₂), 3.24(t, 2H, Lys ϵ -CH ₂), 3.72(s, 3H, OCH ₃),		
			3.79(m, 2H, Ala α -CH×2), 4.05(m, 8H, CH ₂ O×4), 4.53(m, 2H, Ala NH×2), 4.67(t,		
		6.00	1H, Lys α -CH), 7.18(m, 1H, Lys ϵ -NH), 7.41(m, 1H, Lys α -NH)		
Ic	/15.4	6.98	$0.80-1.02(m, 12H, \text{lle } \delta\text{-CH}_3 \times 2, \gamma\text{-CH}_3 \times 2), 1.05-1.20(m, 4H, \text{lle } \gamma\text{-CH}_2 \times 2), 1.28(m, 2)$		
	[M+H]		24H, CH ₃ ×8), 1.30-1.65(m, 4H, Lys γ -CH ₂ , δ -CH ₂), 1.70-2.00(m, 4H, Lys β -CH ₂ ,		
			IIE p-CH×2), $3.24(t, 2H, Lys \epsilon-CH_2)$, $3.40-3.90(m, 4H, IIE \alpha-CH×2, NH×2), 2.71(a, 2H, OCH), 4.20, 4.70(m, 5H, CHO)/4, Lys et CH), 7.17(m, 1H, Lys e, NH).$		
			$3.71(s, 5H, OCH_3), 4.30-4.70(m, 5H, CHOX4, Lys (2-CH), 7.17(m, 1H, Lys e-NH), 7.48(4, 1H, Lys e, NH)$		
1.1	715 2	672	7.40(0, 10, Lys 0, 100)		
10	/13.3	0.75	δ CH β CH and Leu β CH 21 , $1.20(11, 24H, CH_{3}\times 6)$, $1.40-1.95(11, 10H, Lys \gamma-CH2, \delta CH \beta CH and Leu \beta CH 22, 2.51(m, 2H, Leu \alpha CH2), 3.24(t, 2H, Lys)$		
	[[11]]		s_{-CH_2} , p_{-CH_2} and Ecu $p_{-CH_2/2}$, 2.51(iii, 21i, Ecu $p_{-CH_2/2}$), 5.24(i, 21i, Eys $s_{-CH_2/2}$) 3.71(m 2H Leu $\alpha_{-CH_2/2}$) 3.73(s 3H OCH_2) 4.09(m 2H Leu NH22)		
			$4.57(m, 5H, CHOx4, Lys, \alpha, CH)$ 7.28(br s, 1H, Lys, s, NH) 7.41(d, 1H, Lys, \alpha, NH)		
1e	783.2	5.99	$1.26(m, 24H, CH_2 \times 8)$ 1.35-1.55(m, 4H, Lys v-CH ₂ and δ -CH ₂) 1.60-1.85(m, 2H)		
	$[M+H]^+$	0.77	Lys B-CH ₂), 2.9-3.3(m, 6H, Phe B-CH ₂ ×2 and Lys ϵ -CH ₂), 3.71(s, 3H, OCH ₃),		
	. ,		3.65-4.10(m, 4H, Phe α -CH×2, NH×2), 4.25-4.60(m, 5H, CHO×4, Lys α -CH),		
			7.0-7.40(m, 12H, Ph-H×10 and Lys α-NH, ε-NH)		
1f	683.2	6.61	1.31(m, 24H, CH ₃ ×8), 1.45(m, 2H, Lys γ-CH ₂), 1.65-2.05(m, 8H, Lys δ-CH ₂ ,		
	$[M+H]^+$		β-CH ₂ and Pro β-CH ₂ ×2), 2.22(m, 4H, Pro γ-CH ₂ ×2), 3.24(m, 6H, Lys ε-CH ₂ and		
			Pro δ-CH ₂ ×2), 3.72(s, 3H, OCH ₃), 4.14(m, 2H, Pro α-CH×2), 4.57(m, 5H, CHO×4,		
			Lys α-CH), 7.36(t, 1H, Lys ε-NH), 7.74(d, 1H, Lys α-NH)		
1g	663.3	7.16	1.30(m, 24H, CH ₃ ×8), 1.42(m, 2H, Lys γ -CH ₂), 1.67(m, 2H, Lys δ -CH ₂), 1.88(m,		
	$[M+H]^+$		2H, Lys β -CH ₂), 3.11(m, 2H, Lys ϵ -CH ₂), 3.47(m, 2H, OH×2), 3.72(s, 3H, OCH ₃),		
			3.65-3.80(m, 2H, Ser α -CH×2), 4.14(d, 4H, Ser CH ₂ ×2), 4.30(m, 2H, Ser NH×2),		
			4.58(m, 5H, CHO×4, Lys α-CH), 7.21(t, 1H, Lys ε-NH), 7.55(d, 1H, Lys α-NH)		
1h	691.2	7.04	1.32(m, 30H, CH ₃ ×8 and Thr CH ₃ ×2), 1.51(m, 2H, Lys γ-CH ₂), 1.70(m, 2H, Lys		
	[M+H]		δ -CH ₂), 1.86(m, 2H, Lys β -CH ₂), 3.07(m, 2H, OH×2), 3.54(t, 2H, Lys ϵ -CH ₂),		
			$3.65-3.80(m, 2H, 1hr \alpha-CH\times 2), 3./3(s, 3H, OCH_3), 3.85-4.1/(m, 2H, 1hr NH\times 2), 4.40(a) 2H, The R CH-2) A(2)(a) = 5H - CH-3$		
			4.40(m, 2H, Inr p-CH×2), 4.62(m, 5H, CHO×4, Lys α -CH), 7.17(m, 1H, Lys α -NU), 7.52(d, 1H, Lys α -NU)		
1;	697 2	7.04	$0.05(m 124 \text{ Vol} \text{ CH} \times 4)$ 1.29(m 244 CH × 8) 1.45.1.65(m 44 Ly × 64 cH and		
11	1007.3	7.04	δ CH ₂) 1 80(m 2H Lyc B CH ₂) 2 17(m 2H Val B CH ₂) 3 25(t 2H Lyc		
	[[11]]		ϵ -CH ₂), 1.00(m, 2H, Eys p-eH ₂), 2.17(m, 2H, Var p-eH ₂), 3.25(t, 2H, Eys e-CH ₂) 3.45-3.61(m, 2H, Val NHx ²) 3.71(s, 3H, OCH ₂) 3.81(m, 2H, Val		
			α -CH×2) 4 56(m 5H CHO×4 Lys α -CH) 7 20(t 1H Lys ϵ -NH) 7 48(d 1H		
			$I \text{ vs } \alpha \text{-NH}$		
2	616.2	6.29	1.33(m, 24H, CH ₃ ×8), 1.41(m, 8H, Ala CH ₃ ×2 and Lys γ -CH ₂), 1.54(m, 2H, Lys		
	[M+H] ⁺		δ-CH ₂), 1.87(m, 2H, Lys β-CH ₂), 2.87(br s, 1H, Ala NH), 3.26(t, 2H, Lys ε-CH ₂),		
			3.77(m, 2H, Ala α-CH×2), 4.44(m, 1H, Ala NH), 4. 58(m, 5H, CHO×4, Lys		
			α-CH), 6.47(d, 2H, NH ₂), 7.22(m, 1H, Lys ε-NH), 7.57(d, 1H, Lys α-NH)		
3	617^{*}	6.41	1.24-1.32(m, 24H, CH ₃ ×8), 1.39(m, 8H, Lys γ -CH ₂ and Ala CH ₃ ×2), 1.54(m, 2H,		
	$[M+H]^+$		Lys $\beta\text{-}CH_2\text{)},\;1.82(m,\;2H,\;Lys\;\;\delta\text{-}CH_2\text{)},\;3.32(t,\;2H,\;Lys\;\;\epsilon\text{-}CH_2\text{)},\;3.76(m,\;2H,\;Ala$		
			$\alpha\text{-CH}{\times}2),4.08(t,2H,Ala$ NH ${\times}2),4.40(m,1H,Lys$ $\alpha\text{-CH}),4.48(m,4H,CHO{\times}4),$		
			7.22(t, 1H, Lys ε-NH), 7.55(d, 1H, Lys α-NH)		

Table 2MS, ³¹P NMR and ¹H NMR data of compounds 1, 2 and 3

* FAB-MS. ** CDCl3 as solvent.

The preparation of compound **2**: Compound **1a** 0.5 g (0.8 mmol) was dissolved in methanol (5 mL), and 5 mL of 25% ammonia solution was added. The mixture was stirred at room temperature for 4 hours, then evaporated *in vacuo* to remove methanol. 5 mL of water was added to the residue and extracted with CHCl₃ 3 times. The CHCl₃

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extracts were washed with saturated NaCl solution, dried over anhydrous Na_2SO_4 and evaporated *in vacuo*. The residue was recrystallized from THF to give compounds **2** (white solid, 0.3 g).

The preparation of compound **3**: A solution of compound **1a** 0.88 g (1.4 mmol) in methanol (6 mL) was stirred and cooled in an ice-water bath. An aqueous solution of 1 mol/L NaOH (1.7 mL) was added and stirring was continued for 2 hours at room temperature. After removing methanol by evaporation *in vacuo*, 5 mL of water was added and the solution was acidified to pH = 3 with 1 mol/L HCl. A small amount of undissolved substance was filtered off and the filtrate extracted with ethyl acetate 3 times. The combined ethyl acetate extracts was washed with saturated NaCl solution, dried over anhydrous Na₂SO₄ and evaporated *in vacuo*. The residue was recrystallized from ethyl acetate-petroleum ether to yield compound **3** (white solid, 0.62 g).

Tumor growth inhibition of synthesized compounds on solid or non-solid human tumor cell lines has been determined by MTT (3-(4, 5-dimethylthiozol-2-yl)-2, 5-diphenyltetrazolium bromide) tests. The preliminary results showed that percentage of growth inhibition on K562 and A2780 cells of compounds **1d** and **1e** are more than 50% at the concentration of 100 μ g/mL, while that of other compounds are less than 50%. Further MTT tests of compounds **1d** and **1e** at the concentration of 1, 10, 20, 40, 80 and 100 μ g/mL showed that IC₅₀ of compound **1d** on K562 and A2780 cells are 21.37 μ g/mL and 90.41 μ g/mL; and IC₅₀ of compound **1e** on K562 and A2780 cells are 17.44 μ g/mL and 37.82 μ g/mL, respectively. It was observed under microscope that the cell shape exhibited typical characteristic of apoptosis, so it is presumed that the tumor growth inhibition of these two compounds is effected by inducing cell apoptosis. The mechanism and structure-activity relationship remain to be further studied.

Acknowledgment

This work is supported by Natural Science Foundation of Beijing City (7002006). Elemental analyses were performed by Institute of Chemistry, Chinese Academy of Science.

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Received 21 May, 2002